

Antioxidant Activity of Brown Soybean Ethanolic Extracts and Application to Cooked Pork Patties

Choong-Hee Lee, Ko-Eun Hwang, Hyun-Wook Kim¹, Dong-Heon Song, Yong-Jae Kim, Youn-Kyung Ham, Yun-Sang Choi², Sung-Jin Jang, Tae-Jun Jeong, and Cheon-Jei Kim*

Department of Food Science and Biotechnology of Animal Resources, Konkuk University, Seoul 05029, Korea

¹*Meat Science and Muscle Biology Lab, Purdue University, West Lafayette, IN 47907, USA*

²*Food Processing Research Center, Korean Food Research Institute, Seongnam 13539, Korea*

Abstract

The brown soybean extract (BE, extracted by distilled water, 50%, 75%, and 95% ethanol) were analyzed for their total phenol, flavonoid, anthocyanin content, and DPPH radical-scavenging activity to determine antioxidant activities. Brown soybean extract with 75% ethanol showed significantly higher DPPH radical scavenging activity, total phenol and anthocyanin content compared to the other treatments ($p < 0.05$). Then, brown soybean extract with 75% ethanol was applied to pork patties at different concentration (0.05%, 0.1%, and 0.2%) and lipid oxidation was evaluated during 15 d of refrigerated storage. Addition of BE significantly increased redness and pH values, respectively ($p < 0.05$). Moreover, TBARS value of pork patties decreased significantly ($p < 0.05$) as BE concentration increased. In sensory evaluation, pork patties with 0.1% BE had significantly higher score than other treatments in flavor and overall acceptability ($p < 0.05$). Consequently, these results indicate that 0.1% BE could be an effective natural antioxidant to inhibit lipid oxidation in pork patties.

Keywords: brown soybean, anthocyanin, antioxidant, shelf-life, pork patty

Received August 7, 2015; Revised March 21, 2016; Accepted March 24, 2016

Introduction

Lipid oxidation is one of the major causes of quality deterioration in meat products such as off-flavor, discoloration, decrease in nutrition value, and reduction of shelf-life. Delaying lipid oxidation has become a major issue in the meat industry (Fernández-López *et al.*, 2005). Generally, antioxidants have been used to prevent lipid oxidation and maintain qualities of meat (Jia *et al.*, 2012) and synthetic antioxidants such as butylated hydroxytoluene (BHT) and butylated hydroxyanisole (BHA) have been used widely in the food industry. However, replacing synthetic antioxidants with natural antioxidants has been studied because of their potential toxicity (Formanek *et al.*, 2001). Recently, using natural antioxidants from various plant sources have been studied in meat and meat products (Shah *et al.*, 2014).

Soybeans (*Glycine max*) have been regarded as a nutritionally rich food resource. Recently, soybeans have been used as a forage crop and become an issue as functional health foods (Lee *et al.*, 2013). Numerous studies have shown that soybeans contain not only various nutrients such as proteins, unsaturated fatty acids, and carbohydrate, but also abundant bioactive compounds, like vitamins, minerals, isoflavones, saponins, tocopherol, and phytic acid, which have antioxidant activities and synergistic effect with anthocyanins (Myung and Hwang, 2008; Wu *et al.*, 1998).

Some soybean seed coats are colored with red, black, or brown soybeans. Black and brown soybeans are dark colored because of abundant anthocyanins. Anthocyanins are included in red, purple, and blue colored flavonoid pigments in the seed coats of colored soybeans (Choung *et al.*, 2003). Anthocyanins have antioxidant and anti-inflammatory activities without side effect (Myung and Hwang, 2008). Previous research showed that ethanol extract of brown soybeans have high radical-scavenging activity in proportion to amount of total phenolic content and procyanidin (precursor of anthocyanin) (Takahata *et al.*, 2001).

*Corresponding author: Cheon-Jei Kim, Department of Food Science and Biotechnology of Animal Resources, Konkuk University, Seoul 05029, Korea. Tel: 82-2-450-3684, Fax: 82-2-444-6695, E-mail: kimcj@konkuk.ac.kr

Also, Takahashi *et al.* (2005) reported that colored soybeans have antioxidant effect because of anthocyanin contents. Thus, colored beans, including brown soybeans, have more radical-scavenging activity than less colored beans (Alkond *et al.*, 2011). Bae and Moon (1997) reported that brown soybeans had higher antioxidant effects than yellow and black soybeans because of their high contents of phenolic and anthocyanin compounds. Thus, previous investigations have shown that brown soybeans are rich source of various phenolic and anthocyanin compounds. However, brown soybeans have not been researched as natural antioxidants in ethanol extract form to inhibit lipid oxidation in meat products.

Thus, the purpose of this study was to determine the antioxidant activity of ethanol extract of brown soybeans, depending on various ethanol concentration. Furthermore, we investigated the effects of brown soybean ethanol extracts in pork patties on pH, color, thiobarbituric acid reactive substances value, total plate count, and sensory property during refrigerated storage (4°C).

Materials and Methods

Experiment 1. Antioxidant activity of brown soybean extracts (BE) by different levels of ethanol

Preparation of brown soybean extracts (BE)

Brown soybeans were purchased from Yecheon (Korea). Brown soybeans were ground from local market (Korea). Then brown soybean powder (15 g) was extracted with 150 mL of distilled water, 50%, 75%, and 95% ethanol for 24 h in a shaker at room temperature. The extracts was filtered through filter papers (ϕ 110 mm, Cat. No. 1001 110, Whatman International Ltd., England) and evaporated with a rotary evaporator (EYELA N-1000, RIKAKIKAI. Co. Ltd., Japan) below 50°C. After evaporation, brown soybean extracts were dissolved in distilled water, 50, 75, and 95% ethanol.

Extraction yield

Extraction yield for each treatment was calculated by subtracting the dried weight of plant material residue after extraction from the weight of the original plant material.

Total phenolic content

The total phenol content was determined according to the Folin-Ciocalteu method of Slinkard and Singleton (1977). An aliquot of each extract was mixed with 4.5 mL of distilled water. Then 0.1 mL of Folin-Ciocalteu reagent

(previously diluted 3-fold with distilled water) was added and mixed using a vortex mixer. After 5 min, 0.3 mL of 2% sodium carbonate solution was added. The mixture was allowed to stand for 2 h. The absorbance was measured at 760 nm.

Total flavonoid content

The total flavonoid content of the samples was determined by a modified colorimetric method described by Myung and Hwang (2008) using quercetin as a standard. Extracts 1 mL were mixed with 4 mL of distilled water and 0.3 mL of 5% sodium nitrite (NaNO₂) solution followed by the addition of 0.3 mL of 10% aluminum chloride (AlCl₃) solution 5 min later. After 2 mL of 1 M sodium hydroxide (NaOH) and 2.4 mL distilled water was added. The solutions were mixed and absorbance was measured at 510 nm. The results were expressed in mg quercetin/g extract.

Total anthocyanin content

Determination of total anthocyanin component was measured according to a modification of the pH-differential methods (Hosseinian, 2008). Anthocyanin is reversible structural transformations with a change in pH. The oxonium (orange-purple) form predominates at pH 1.0 and the hemiketal (colorless) form at pH 4.5. The pH-differential method is based on this reaction, and rapid measurements for the total amount of anthocyanin. Sample (0.2 mL) was added to 1.8 mL of two buffers (one for pH 1.0 using potassium chloride buffer and the other for pH 4.5 using sodium acetate buffer). The absorbance of each sample was measured 520 nm and 700 nm against distilled water as blank. The concentration (mg/L) of each anthocyanin was calculated according to the following formula:

$$\begin{aligned} \text{Total anthocyanin content (mg/L)} \\ = A \times MW \times DF \times 10^3 / \epsilon \times 1 \end{aligned}$$

$$A \text{ (absorbance value)} = (A_{520 \text{ nm}} - A_{700 \text{ nm}})_{\text{pH } 1.0} - (A_{520 \text{ nm}} - A_{700 \text{ nm}})_{\text{pH } 4.5}$$

$$MW \text{ (molecular weight of cyanidin-3-glucoside)} = 449.2$$

$$DF \text{ (dilution factor)} = \text{dilution ratio of sample}$$

$$\epsilon \text{ (cyanidin-3-glucoside molar absorbance)} = 26,900 \text{ M}^{-1} \text{ cm}^{-1}$$

1,1-Diphenyl-2-picrylhydrazyl (DPPH) radical scavenging activity

The free radical scavenging activity of the brown soybean extract was measured using the modified method of Brand-Williams *et al.* (1995). Briefly, 1 mL of extracts at

different concentrations (0.25-0.75 mg/mL) was mixed with 1 mL of a 2×10^{-4} M ethanolic DPPH solution. The reaction mixture was incubated in the dark for 20 min and the absorbance was measured at 517 nm with a UV-Vis spectrophotometer (Libra S22, Biochrom Ltd., UK). Percent of DPPH-scavenging activity was calculated according to the following equation:

$$\text{Scavenging activity (\%)} = [(\text{control absorbance} - \text{extract absorbance}) / (\text{control absorbance})] \times 100$$

Experiment 2. Effect of brown soybean extract on shelf life of pork patties during refrigerated storage for 15 days

Preparation of pork patties with brown soybean extract

Fresh pork hams and back fat were purchased from local processor 48 h postmortem. Visible fat and connective tissues were trimmed off. Lean meat and fat were ground through ϕ -8 mm plate using meat grinder (PM-70, Mainca, Spain). The ground meat (65%) and fat (25%) were mixed in the bowl with water (10%), salt (1.5%), and then ascorbic acid (ASC) or brown soybean extract with 75% ethanol (pH, 6.22 ± 0.02 ; L*-value, 40.05 ± 2.39 ; a*-value, 24.88 ± 1.09 ; b*-value, 25.32 ± 0.84) were added according to the following formulation: CON (no antioxidant added); ASC-0.05 (0.05% ascorbic acid); BE-0.05 (0.05% brown soybean extract with 75% ethanol); BE-0.1 (0.1% brown soybean extract with 75% ethanol); BE-0.2 (0.2% brown soybean extract with 75% ethanol). Each mixed sample was divided into 5 portions and allocated to the 5 periods (0, 3, 7, 10 and 15 d). Samples weighed in petri dish (about 85 g) and cooked using water bath (VS-190W, Vision Scientific Co., Korea) until the core temperature reached 75°C for 30 min (Du *et al.*, 2001). The cooked pork patties were vacuum-packed in PE/Nylon film bags, stored at $4 \pm 1^\circ\text{C}$ for 15 d.

pH measurement

The pH values of cooked pork patties were determined with a pH meter (Model 340, Mettler-Toledo GmbH, Switzerland). The pH values of samples were measured by blending a 5 g sample with 20 mL distilled water for 30 s in a homogenizer at 8,000 rpm (Ultra-Turrax SK15, Janke & Kunkel, Germany).

Color measurement

Instrumental color were determined using a colorimeter

(Minolta Chroma meter CR-210, Japan; illuminate C, calibrated with a white plate, CIE L* \approx +97.83, CIE a* \approx -0.43, CIE b* \approx +1.98). Five measurements for five locations on surface of cooked pork patties were taken. CIE L* (lightness), CIE a* (redness), and CIE b* (yellowness) values were recorded.

Thiobarbituric acid reactive substances (TBARS) value

Lipid oxidation was assessed in triplicate using the TBARS method of Tarladgis *et al.* (1960) with minor modifications. A 10 g sample was blended with 50 mL of distilled water and 0.2 mL of butylated hydroxyl toluene (BHT) for 2 min and then transferred to a distillation tube. The cup used for blending was washed with an additional 47.5 mL of distilled water, which was added to the same distillation flask with 2.5 mL of 4 N HCl and a few drops of a silicone o/w antifoam agent (KMK-73, Shin-Etsu Silicone Co. Ltd., Korea). The mixture was distilled and 5 mL of the distillate was mixed with 0.02 M 2-thiobarbituric acid in 90% acetic acid (TBA reagent) in a vial. The vial were capped and heated in boiling water bath for 35 min to develop the chromogen and then cooled to room temperature. Absorbance was measured using a UV/VIS spectrophotometer (Optizen 2120 UV plus, Mecasys Co. Ltd., Korea) at 538 nm, against a blank prepared with 5 mL distilled water and 5 mL TBA-reagent. The TBA values were calculated as malondiadehyde (MDA) per kilogram of sample (mg MDA/kg meat). The formula was:

$$\text{TBARS (MDA mg/meat kg)} = (\text{optical density of sample} - \text{optical density of blank}) \times 7.8$$

Total plate count

To determine the total aerobic bacterial count for each treatment, 10 g of the sample was aseptically transferred into a sterile stomacher bag and 90 mL of sterile 0.1% peptone water (Difco, USA) was added. The sample was then homogenized in stomacher (Masticater-Paddle-Blender, IUL Instrument, Spain) for 2 min at normal speed and serial dilutions of the homogenate were prepared with 0.1% peptone water. After serially diluting each sample, 0.1 mL aliquots were spread onto plates respectively. The total bacterial count was determined on plate count agar (PCA, Difco, USA) at 37°C for 48 h. Microbial colonies were counted and expressed as Log CFU/g sample of pork patty.

Sensory evaluations

Sensory evaluations were performed in duplicate on each

pork patty by panelist. The samples were cut into quarters and served to the panels when samples were at room temperature. The pork patties were evaluated for color (1=extremely undesirable, 10=extremely desirable), flavor (1=extremely undesirable, 10=extremely desirable), off-flavor (1=extremely undesirable, 10=extremely desirable), and overall acceptability (1=extremely undesirable, 10=extremely desirable) using a 10 point descriptive scale. The panel consisted of 10 members from the Department of Food Sciences and Biotechnology of Animal Resources at Konkuk University in Korea.

Statistical analysis

All tests were done at least three times for each experimental condition and mean values were reported. The statistical analysis of all data was performed by SPSS Ver. 18.0 (SPSS Inc., USA). One-way ANOVA (one-way analysis of variance) and Duncan's multiple range comparison were used to find the level of significant differences ($p < 0.05$).

Results and Discussion

Experiment 1. Antioxidant activity of brown soybean extracts with different levels of ethanol

Extraction yield

Extraction yield of brown soybean extracts with different levels of ethanol are shown in Fig. 1. The extraction yield of BE decreased as ethanol concentration increased. The 95% EtOH showed the lowest extraction yield ($p < 0.05$). DW and 50% EtOH treatments showed significantly ($p < 0.05$) higher extraction yields than 75% EtOH. Similarly, Lee *et al.* (2014) reported that extraction yield increased as ethanol concentration decreased. Kim *et al.* (1993) suggested that natural antioxidant extracted by diluted ethanol solution has higher extraction yield compared to that which was extracted by pure ethanol solution. Because antioxidant compounds in plant have different polarity and solubility (Sun and Ho, 2005), extraction solvent property may affect the extraction yield. Cacace and Mazza (2003) reported that different concentration of ethanol affects the physical properties of the solvent, and this may change the extraction yield of various bioactive compounds. Nonpolar covalent molecules can be dissolved less in low concentration ethanol solvent because the energy needed to breakdown the interaction between the water molecules.

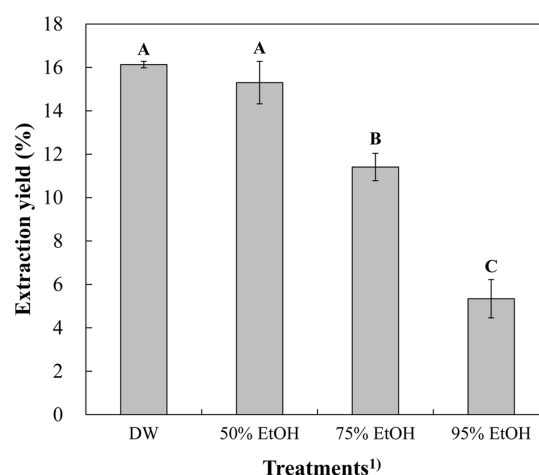


Fig. 1. Extraction yield of brown soybean extracts by different levels of ethanol. ¹⁾Treatments: DW, brown soybean extracted with distilled water; 50% EtOH, brown soybean extracted with 50% ethanol; 75% EtOH, brown soybean extracted with 75% ethanol; 95% EtOH, brown soybean extracted with 95% ethanol. The vertical bars represent mean±standard deviation. ^{A-C}Mean sharing different letters are significantly different ($p < 0.05$).

Total phenol and flavonoid contents

The total phenol and flavonoid contents of BE extracted by different levels of ethanol are shown in Table 1. Total phenol content showed significant differences between the treatments. Especially, the 75% EtOH extract had significantly higher total phenolic contents than the other treatments ($p < 0.05$). Lee *et al.* (2010) showed similar results that total phenol contents in ethanol extract of large deltooid sunurus (*Synurus excelsus*) and Chrysanthemum-leaf synurus (*Synurus palmatopinnatifidus*) were higher than those of distilled water extract.

Flavonoids, including flavones, flavanones, isoflavones,

Table 1. Total phenol and flavonoid contents of brown soybean extracts by different levels of ethanol

Treatment ¹⁾	Total phenol content (mg of gallic acid equivalents/g)	Total flavonoid content (mg of quercetin equivalents/g)
DW	15.04±5.48 ^D	5.85±3.97 ^C
50% EtOH	28.11±5.87 ^C	4.33±2.40 ^C
75% EtOH	54.17±4.77 ^A	7.40±2.54 ^B
95% EtOH	41.93±2.36 ^B	12.93±4.53 ^A

All values are mean±standard deviation.

¹⁾Treatments: DW, brown soybean extracted with distilled water; 50% EtOH, brown soybean extracted with 50% ethanol; 75% EtOH, brown soybean extracted with 75% ethanol; 95% EtOH, brown soybean extracted with 95% ethanol.

^{A-D}Mean sharing different letters are significantly different ($p < 0.05$).

flavonols and anthocyanidins, have antioxidant activity in food system. In particular, legumes, such as soybean, colored beans, and chick peas contain abundant flavonoids (Pietta, 2000). Total flavonoid contents of the 95% EtOH were significantly higher than the other treatments ($p<0.05$), followed by 75%, 50% EtOH and DW. Total flavonoid contents were higher in ethanolic extract than distilled water extract, similarly to the total phenol contents.

Total anthocyanin content

Total anthocyanin contents in BE extracted by different levels of ethanol are presented in Fig. 2. The 75% EtOH showed the highest content of anthocyanin ($p<0.05$), followed by 50% EtOH, DW and 95% EtOH. According to Kim *et al.* (2014), 60-75% ethanol was more effective extract solvent than other concentrations of ethanol. These differences were reported that anthocyanins were influenced by chemical structure and pH (Choung *et al.*, 2008). Also, polymerized anthocyanin and non-enzymatic browning pigments were excluded from the calculation because these compounds do not exhibit reversible behavior with pH (Wrolstad *et al.*, 2005). Similarly, Kim *et al.* (2008) suggested that the efficient extraction solvent for anthocyanin was 60-80% methanol.

1,1-Diphenyl-2-picrylhydrazyl (DPPH) radical scavenging activity

DPPH radical scavenging activity of BE are shown in Fig. 3. As extract solvent (ethanol) concentration increased,

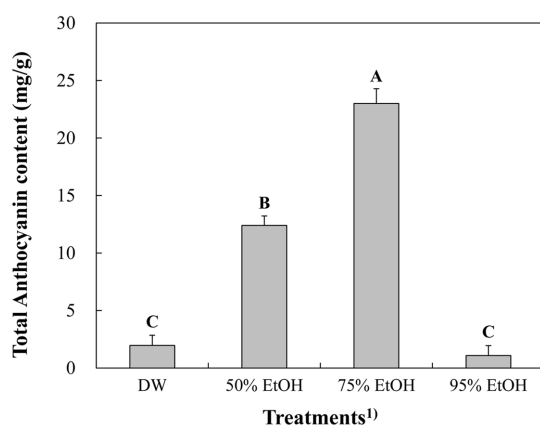


Fig. 2. Total anthocyanin contents in brown soybean extracts by different levels of ethanol. ¹⁾Treatments: DW, brown soybean extracted with distilled water; 50% EtOH, brown soybean extracted with 50% ethanol; 75% EtOH, brown soybean extracted with 75% ethanol; 95% EtOH, brown soybean extracted with 95% ethanol. The vertical bars represent mean±standard deviation. ^{A-C}Mean sharing different letters are significantly different ($p<0.05$).

DPPH radical-scavenging activity increased and the BE extracted by 75% ethanol showed the highest DPPH radical-scavenging activity among the treatments ($p<0.05$). Jun *et al.* (2014) reported that a mixed solvent was more efficient for extracting antioxidant compounds compared to a pure solvent. Additionally, natural antioxidant compounds, such as isoflavones and phenol compounds, are extracted better by 70-80% ethanol (Bae *et al.*, 1997). In this study, 75% ethanol is considered as the most effective solvent to extract antioxidant compounds. The most amount of anthocyanin and total phenol contents in 75% EtOH may have accounted for the highest DPPH radical-scavenging activity. The result of anthocyanin content analysis showed that 75% EtOH presented the highest amount of anthocyanin among the treatments (Fig. 2). In previous study, the high content of anthocyanin in soybeans was shown to possess strong antioxidant activity (Alkond *et al.*, 2011). Kim *et al.* (2009) reported that DPPH radical-scavenging activity was correlated with polyphenol content which is abundant in soybean. Phenolic compounds act as a reductant reacting with free radicals to make stable products and terminate free radical chain reactions (Neci and Jayaprakasha, 2003).

Experiment 1 showed that brown soybean extract with 75% ethanol has the highest antioxidant activity. The high content of anthocyanin and phenol compounds in 75% EtOH were assumed to play an important role in scavenging

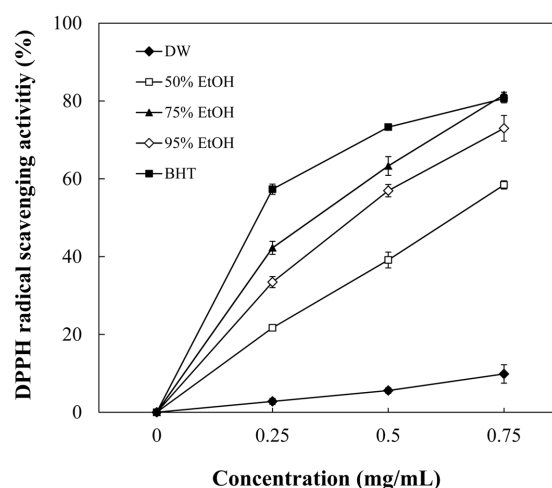


Fig. 3. Antioxidant activity of brown soybean extracts, measured as percent scavenging of DPPH radical. ¹⁾Treatments: DW, brown soybean extracted with distilled water; 50% EtOH, brown soybean extracted with 50% ethanol; 75% EtOH, brown soybean extracted with 75% ethanol; 95% EtOH, brown soybean extracted with 95% ethanol; BHT, synthetic antioxidant (butylatedhydroxytoluene). Each bar represents a mean±standard deviation.

ing free radicals. Thus, addition of brown soybean extract with 75% ethanol was expected to be effective for retard lipid oxidation as a natural antioxidant on pork patties.

Experiment 2. Effect of brown soybean extract (BE) on shelf-life of pork patties during refrigerated storage for 15 days

pH of pork patties with BE

The pH values of pork patties with BE during refrigerated storage are shown in Table 2. The pH values of pork patties were decreased significantly as storage time increased ($p<0.05$). Similar results were found by Song *et al.*

(1997), who revealed that dissociation of CO_2 in muscle tissue can cause decreases in pH during storage. The pH values of pork patties showed significant differences with addition of BE or ascorbic acid. The pH values of pork patties with 0.2% BE were significantly higher than the other treatments during refrigerated storage ($p<0.05$). This result may be attributed to the pH of brown soybean extract (6.22 ± 0.02).

Color evaluation of pork patties with BE

The results of color measurements of pork patties formulated with BE during 15 d storage are shown in Table 3. Lightness (CIE L^* -value) of pork patties with BE was

Table 2. Change in pH values of pork patties with added brown soybean extract during refrigerated storage for 15 d

Traits	Storage periods (d)	CON	Treatments ¹⁾			
			ASC-0.05	BE-0.05	BE-0.1	BE-0.2
pH	0	6.17±0.03 ^{Ab}	6.12±0.02 ^{Ac}	6.17±0.01 ^{Ab}	6.18±0.01 ^{Ab}	6.22±0.01 ^{Aa}
	3	6.08±0.01 ^{Bc}	6.04±0.01 ^{Bd}	6.07±0.01 ^{Bc}	6.10±0.02 ^{Bb}	6.13±0.01 ^{Ba}
	7	6.01±0.02 ^{Cc}	6.02±0.01 ^{Bc}	6.07±0.01 ^{Bb}	6.11±0.02 ^{Ba}	6.12±0.01 ^{Ca}
	10	5.98±0.01 ^{Cb}	5.93±0.01 ^{Cc}	6.01±0.01 ^{Ca}	5.98±0.01 ^{Cb}	6.02±0.01 ^{Da}
	15	5.89±0.05 ^{Db}	5.90±0.02 ^{Db}	5.95±0.01 ^{Da}	5.98±0.01 ^{Ca}	5.99±0.01 ^{Ea}

All values are mean±standard deviation.

¹⁾Treatments: CON, pork patty without brown soybean extract; ASC-0.05, pork patty with added 0.05% ascorbic acid; BE-0.05, pork patty with added 0.05% brown soybean extract; BE-0.1, pork patty with added 0.1% brown soybean extract; BE-0.2, pork patty with added 0.2% brown soybean extract.

^{A-E}Means within a column with different letters are significantly different ($p<0.05$).

^{a-d}Means within a row with different letters are significantly different ($p<0.05$).

Table 3. Change in color of pork patties with added brown soybean extract during refrigerated storage for 15 d

Traits ²⁾	Storage periods (d)	CON	Treatments ¹⁾			
			ASC-0.05	BE-0.05	BE-0.1	BE-0.2
CIE L^*	0	66.37±0.96 ^{Aa}	66.47±1.11 ^{Aa}	65.96±0.39 ^{Aab}	66.24±0.58 ^{Aa}	65.30±0.62 ^{Ab}
	3	65.22±0.57 ^{Ba}	63.72±1.05 ^{Cbc}	64.09±0.69 ^{Bb}	63.11±0.95 ^{Ccd}	62.93±0.78 ^{Cd}
	7	65.80±0.80 ^{Ba}	65.70±0.72 ^{ABa}	63.68±0.84 ^{Bbc}	63.88±0.94 ^{BCb}	62.97±1.36 ^{Cc}
	10	63.89±0.99 ^{Ca}	65.20±0.51 ^{Bc}	63.93±1.06 ^{Bb}	64.09±0.94 ^{Cc}	64.40±0.77 ^{Bd}
	15	65.33±0.89 ^{Ba}	64.90±1.60 ^{Bb}	63.56±0.82 ^{Bb}	63.44±0.96 ^{BCd}	63.22±0.70 ^{Cc}
CIE a^*	0	5.07±0.28 ^{Aab}	4.76±0.27 ^{Ac}	5.02±0.16 ^{BCab}	4.91±0.12 ^{Cbc}	5.22±0.19 ^{BCa}
	3	5.07±0.16 ^{Aa}	4.60±0.13 ^{ABb}	4.94±0.34 ^{Ca}	4.98±0.19 ^{Ca}	5.08±0.23 ^{Ca}
	7	5.06±0.32 ^{Ac}	4.46±0.26 ^{Bd}	5.21±0.17 ^{ABbc}	5.44±0.29 ^{Aab}	5.59±0.23 ^{Aa}
	10	5.06±0.21 ^{Ab}	4.75±0.23 ^{Ac}	5.38±0.26 ^{Aab}	5.21±0.34 ^{Bab}	5.40±0.46 ^{ABa}
	15	5.20±0.22 ^{Aa}	4.54±0.28 ^{ABb}	5.33±0.26 ^{Aa}	5.36±0.12 ^{ABa}	5.32±0.29 ^{ABCa}
CIE b^*	0	11.79±0.51 ^{Ac}	12.41±0.28 ^{Aa}	12.25±0.19 ^{Aab}	12.28±0.35 ^{Aab}	12.02±0.16 ^{Bbc}
	3	11.47±0.47 ^{Bb}	11.89±0.39 ^{Ba}	11.85±0.32 ^{Ba}	11.81±0.27 ^{Ba}	11.83±0.31 ^{Ba}
	7	11.51±0.55 ^{Ac}	12.53±0.28 ^{Aa}	12.08±0.34 ^{ABb}	11.88±0.47 ^{Bbc}	10.77±0.44 ^{Cd}
	10	12.39±0.31 ^{Aa}	12.51±0.37 ^{Aa}	12.10±0.50 ^{ABab}	11.72±0.62 ^{Bb}	12.33±0.13 ^{Aa}
	15	11.84±0.32 ^{Accd}	12.64±0.36 ^{Aa}	12.08±0.21 ^{ABbc}	12.29±0.35 ^{Ab}	11.73±0.37 ^{Bd}

All values are mean±standard deviation.

¹⁾Treatments: CON, pork patty without brown soybean extract; ASC-0.05, pork patty with added 0.05% ascorbic acid; BE-0.05, pork patty with added 0.05% brown soybean extract; BE-0.1, pork patty with added 0.1% brown soybean extract; BE-0.2, pork patty with added 0.2% brown soybean extract.

²⁾Traits: CIE L^* , Lightness; CIE a^* , redness; CIE b^* , yellowness.

^{A-C}Means within a column with different letters are significantly different ($p<0.05$).

^{a-d}Means within a row with different letters are significantly different ($p<0.05$).

significantly lower than CON and ASC-0.05 treatments ($p < 0.05$). The redness (CIE a^* -value) of sample that BE added was significantly higher than with ASC-0.05 ($p < 0.05$). Also, the lightness of the pork patties decreased during refrigerated storage for 15 d, whereas the redness increased. These results may be due to red-brown color of BE. Similarly, plum extracts increased redness and decreased lightness in turkey breast during storage because of the original dark purple color (Lee and Ahn, 2005). The yellowness (CIE b^* -value) of pork patties was not affected by BE. Similarly, Lee *et al.* (2009) reported that the lightness of anthocyanin in black soybeans was decreased and the yellowness was almost unchanged as heating temperature and time increased. Chen *et al.* (1999) reported that the addition of colored natural antioxidant affected the meat color. These results showed that anthocyanins in BE affected the color of cooked pork patties during refrigerated storage.

TBARS values of pork patties with BE

The results of TBARS analysis to determine lipid oxidation of pork patties for 0, 3, 7, 10, and 15 d are shown in Fig. 4. The TBARS values of pork patties with BE were considerably lower than CON during storage ($p < 0.05$), except BE-0.05 ($p > 0.05$). This result indicates that BE above 0.05% inhibited lipid oxidation in pork patties during storage period. The ability of BE to delay lipid oxidation must be related to high free radical scavenging activity on account of abundant total phenols, flavonoids, and anthocyanins contents. In accordance with the result, previous studies reported that extracts of colored soybeans have high DPPH radical-scavenging activity due to pro-cyanidins and it delayed the formation of MDA in meat products (Jayawardana *et al.*, 2011; Takahata *et al.*, 2001). Besides, BE possess various bioactive compounds such as isoflavones, tocopherols, phenolic acids, and anthocyanins (Kim *et al.*, 2005). According to Naveena *et al.* (2008), anthocyanins and phenolic compounds in pomegranate extract significantly reduced TBARS values of chicken patties during refrigerated storage. Among the treatments, ASC-0.05 showed the lowest TBARS values during refrigerated storage ($p < 0.05$). It has been reported that ascorbic acid inhibited lipid oxidation by reducing superoxide, H_2O_2 , and hydroxyl radicals (Sarma *et al.*, 1997). The TBARS values of all treatments increased at day 3, then began to decrease over the storage period. Min and Ahn (2005) noted that lipid oxidation could be accelerated by any process such as grinding, mincing, and cooking at the beginning of storage. Increase of TBARS values at day 3

may be related to the initial processing of pork patties. It is assumed that decline of TBARS values of samples after day 3 was attributed to a decomposition of malondialdehyde (MDA). Fernandez *et al.* (1997) revealed that MDA and other short-chain carbon products of lipid oxidation are not stable because these products can be decomposed to organic alcohols and acids during storage.

Total plate count of pork patties with BE

Total plate count of pork patties was not detected (< 1 Log CFU/g) during refrigerated storage (data not shown). Cold storage, cooking process, and vacuum-packaging are assumed that have inhibited a growth of aerobic bacteria in the samples. According to Biswas *et al.* (2004), changes in storage temperature reduce the survival rate of microbes, and cooking process drastically injures and/or kills psychrotrophic bacteria. Also, vacuum packaging inhibits microbial growth in meat more than other packaging methods (Seydim *et al.*, 2006). Additionally, Viskelis *et al.* (2009) reported that anthocyanin has antimicrobial activity against *Bacillus cereus*, *Micrococcus luteus*, and *Listeria monocytogenes* in food.

Sensory evaluation of pork patties with BE

Sensory properties of pork patties with BE are shown in Table 4. Regarding color, the pork patties with BE treat-

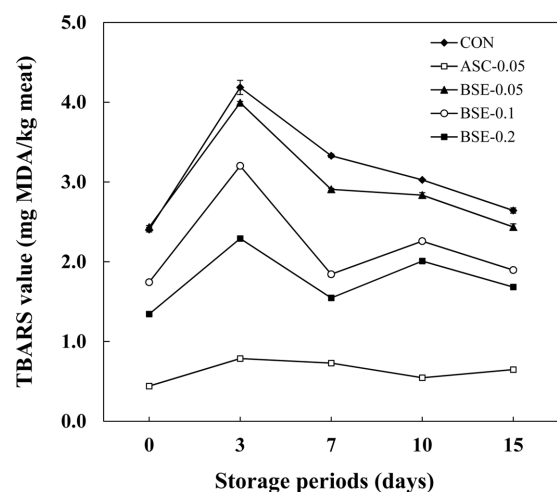


Fig. 4. Change in thiobarbituric acid reactive substances (TBARS) values of pork patties with added brown soybean extract during refrigerated storage for 15 d. ¹⁾Treatments: CON, pork patty without brown soybean extract; ASC-0.05, pork patty with added 0.05% ascorbic acid; BE-0.05, pork patty with added 0.05% brown soybean extract; BE-0.1, pork patty with added 0.1% brown soybean extract; BE-0.2, pork patty with added 0.2% brown soybean extract. Each bar represents a mean \pm standard deviation.

Table 4. Change in sensory evaluation of pork patties with added brown soybean extract during refrigerated storage for 15 d

Traits ²⁾	Storage periods (d)	CON	Treatments ¹⁾			
			ASC-0.05	BE-0.05	BE-0.1	BE-0.2
Color	0	7.40±0.52 ^{Aa}	7.50±0.53 ^{Aa}	7.60±0.52 ^{Aa}	7.70±0.48 ^{Aa}	7.80±0.42 ^{Aa}
	3	7.20±0.63 ^{ABa}	7.10±0.57 ^{ABa}	7.40±0.70 ^{Aa}	7.50±0.71 ^{Aa}	7.60±0.70 ^{Aa}
	7	6.70±0.67 ^{ABCc}	6.80±0.79 ^{Bbc}	7.30±0.67 ^{ABabc}	7.40±0.52 ^{Aab}	7.50±0.53 ^{Aa}
	10	6.60±0.97 ^{BCb}	6.50±0.85 ^{BCb}	6.70±0.82 ^{BCab}	7.40±0.84 ^{Aab}	7.40±0.84 ^{Aa}
	15	6.00±0.94 ^{Cb}	6.00±0.67 ^{Cb}	6.20±0.79 ^{Cab}	6.60±1.07 ^{Bab}	7.00±1.25 ^{Aa}
Flavor	0	7.00±1.15 ^{Aab}	7.50±1.58 ^{Aa}	7.10±1.10 ^{Aa}	6.70±0.82 ^{Aab}	6.00±0.67 ^{Ab}
	3	6.80±0.92 ^{Aab}	7.50±1.08 ^{Aa}	6.60±0.84 ^{Abc}	6.30±0.67 ^{Abc}	5.90±0.57 ^{Ac}
	7	6.90±1.20 ^{Aab}	7.30±1.34 ^{Aa}	6.50±1.27 ^{Aab}	6.20±1.03 ^{Aab}	5.90±1.20 ^{Ab}
	10	6.60±1.07 ^{Aa}	7.10±0.74 ^{Aa}	6.60±0.84 ^{Aa}	6.50±0.71 ^{Aab}	5.70±1.06 ^{Ab}
	15	6.00±1.25 ^{Aab}	6.80±0.92 ^{Aa}	6.50±0.97 ^{Aa}	6.40±0.52 ^{Aab}	5.50±1.08 ^{Ab}
Off-flavor	0	6.10±0.32 ^{Ac}	7.30±1.06 ^{Aa}	6.50±0.84 ^{Abc}	6.80±0.63 ^{Aabc}	7.10±0.74 ^{Aab}
	3	6.10±0.57 ^{Ab}	7.20±0.79 ^{Aa}	6.50±0.71 ^{Aab}	6.30±1.34 ^{ABab}	6.30±1.06 ^{Bab}
	7	6.20±1.75 ^{Aa}	6.70±1.57 ^{ABa}	6.10±0.88 ^{ABa}	6.10±0.74 ^{ABa}	6.30±0.67 ^{Ba}
	10	5.50±0.85 ^{Ab}	6.50±0.85 ^{ABa}	5.60±0.52 ^{Bb}	5.80±1.14 ^{Bab}	5.90±0.88 ^{Bab}
	15	4.50±0.53 ^{Bc}	6.20±0.42 ^{Ba}	5.50±0.53 ^{Bb}	5.70±0.67 ^{Bab}	5.80±0.63 ^{Bab}
Overall acceptability	0	7.00±1.25 ^{Aa}	7.50±1.35 ^{Aa}	7.10±0.88 ^{Aa}	6.70±0.95 ^{Aa}	6.10±0.88 ^{Aa}
	3	6.40±0.84 ^{ABab}	7.20±1.32 ^{Aa}	6.60±0.52 ^{ABab}	6.50±0.97 ^{Aab}	5.80±0.79 ^{Ab}
	7	5.90±0.99 ^{Bb}	7.10±1.37 ^{Aa}	6.30±0.48 ^{BCab}	6.30±1.06 ^{Aab}	5.40±0.97 ^{Ab}
	10	5.80±1.03 ^{Bbc}	6.90±0.74 ^{Aa}	5.70±0.95 ^{CDbc}	6.40±0.70 ^{Aab}	5.50±0.71 ^{Ac}
	15	5.60±1.07 ^{Bbc}	6.80±0.79 ^{Aa}	5.60±0.70 ^{Dbc}	6.30±0.67 ^{Aab}	5.40±1.07 ^{Ac}

All values are mean±standard deviation.

¹⁾Treatments: CON, pork patty without brown soybean extract; ASC-0.05, pork patty with added 0.05% ascorbic acid; BE-0.05, pork patty with added 0.05% brown soybean extract; BE-0.1, pork patty with added 0.1% brown soybean extract; BE-0.2, pork patty with added 0.2% brown soybean extract.

²⁾Traits: color (1=extremely undesirable, 10=extremely desirable), flavor (1=extremely undesirable, 10=extremely desirable), off-flavor (1=extremely undesirable, 10=extremely desirable), and overall acceptability (1=extremely undesirable, 10=extremely desirable).

^{A-D}Means within a column with different letters are significantly different ($p<0.05$).

^{a-c}Means within a row with different letters are significantly different ($p<0.05$).

ments received significantly higher score than CON ($p<0.05$). This result was probably due to red-brown color of BE added to pork patties. In addition, color of pork patties decreased significantly during storage except BE-0.2 ($p<0.05$). The flavor score of BE-0.2 was significantly lower than other treatments ($p<0.05$). Youn *et al.* (2014) reported that increasing an addition level of agent may cause lower score of flavor. The flavor of the pork patties was not changed apparently during refrigerated storage. The off-flavor score of the pork patties showed significant differences between the treatments. Pork patties with antioxidant had significantly higher off-flavor scores than pork patties without antioxidant during refrigerated storage ($p<0.05$). This result indicates that ascorbic acid and brown soybean extract which were added into pork patties inhibited off-flavor generation by delaying lipid oxidation. The overall acceptability score of the pork patties differed significantly among the treatments ($p<0.05$). ASC-0.05 and BE-0.1 had significantly higher overall acceptability score than the other treatments during storage period ($p<0.05$). These results were agreed with Resurreccion (2003) that

reported the appearance and flavor influenced consumer overall acceptability of meat.

Conclusion

In the present study, 75% EtOH had significantly higher DPPH radical-scavenging activity than other gradient ethanol solvent. Also, 75% EtOH had higher estimated contents of phenols and anthocyanins. Addition of BE inhibited lipid oxidation throughout refrigerated storage and affected the redness of pork patties. These results suggest that 0.1% BE can be used as a natural antioxidant that improves shelf-life of meat and meat products during storage.

Acknowledgements

This study was supported by the Brain Korean 21 Plus (BK21 Plus) Project from the Ministry of Education and Human Resources Development (Republic of Korea).

References

- Alkond, A. S. M. G. M., Khandaker, L., Berthold, J., Gates, L., Peters, K., DeLong, H., and Hossain, K. (2011) Anthocyanin, total polyphenols and antioxidant activity of common bean. *Am. J. Food Technol.* **6**, 385-394.
- Bae, E. A., Kwon, T. W., and Moon, G. S. (1997) Isoflavone contents and antioxidative effects of soybeans, soybean curd and their by-products. *J. Korean Soc. Food Sci. Nutr.* **26**, 371-375.
- Bae, E. A. and Moon, G. S. (1997) A study on the antioxidative activities of Korean soybeans. *J. Korean Soc. Food Sci. Nutr.* **26**, 203-208.
- Biswas, A. K., Keshri, R. C., and Bisht, G. S. (2004) Effect of enrobing and antioxidants on quality characteristics of pre-cooked pork patties under chilled and frozen storage conditions. *Meat Sci.* **66**, 733-741.
- Brand-Williams, W., Cuvelier, M. E., and Berset, C. (1995) Use of a free radical method to evaluated antioxidant activity. *Food Sci. Technol.* **28**, 25-30.
- Cacace, J. E. and Mazza, G. (2003) Optimization of extraction of anthocyanins from black currants with aqueous ethanol. *Food Eng. Phys. Properties* **68**, 240-248.
- Chen, X., Jo, C., Lee, J. I., and Ahn, D. U. (1999) Lipid oxidation, volatiles and color changes of irradiated pork patties as affected by antioxidants. *J. Food Sci.* **49**, 429-434.
- Choung, M. G., Choi, B. R., An, Y. N., Chu, Y. H., and Cho, Y. S. (2003) Anthocyanin profile of Korea cultivated kidney bean (*Phaseolus vulgaris* L.). *J. Agric. Food Chem.* **51**, 7040-7043.
- Choung, M. G., Hwang, Y. S., Lee, H. J., Choi, S. S., Lim, J. D., Kang, S. T., Han, W. Y., Baek, I. Y., and Kim, H. K. (2008) Optimal extraction condition of anthocyanins in soybean (*Glycine max*) with black seed coats. *Korean J. Crop. Sci.* **53**, 110-117.
- Du, M., Ahn, D. U., Nam, K. C., and Sell, J. L. (2001) Volatile profiles and lipid oxidation of irradiated cooked chicken meat from laying hens fed diets containing conjugated linoleic acid. *Poultry Sci.* **80**, 235-241.
- Fernández, J., Pérez-Álvarez, J. A., and Fernández-López, J. A. (1997) Thiobarbituric acid test for monitoring lipid oxidation in meat. *Food Chem.* **59**, 345-353.
- Fernández-López, J., Zhi, N., Aleson-Carbonell, L., Pérez-Álvarez, J. A., and Kuri, V. (2005) Antioxidant and antibacterial activities of natural extracts application in beef meatballs. *Meat Sci.* **69**, 371-380.
- Formanek, Z., Kerry, J. P., Higgins, F. M., Buckley, D. J., Morrissey, P. A., and Farkas, J. (2001) Addition of synthetic and natural antioxidants to α -tocopherol acetate supplemented. *Meat Sci.* **58**, 337-341.
- Hosseini, F. S., Li, W., and Beta, T. (2008) Measurement of anthocyanins and other phytochemicals in purple wheat. *Food Chem.* **109**, 916-924.
- Jayawardana, B. C., Hirano, T., Han, K. H., Ishii, H., Okada, T., Shibayama, S., Fukushima, M., Sekikawa, M., and Simada, K. I. (2011) Utilization of adzuki bean extract as a natural antioxidant in cured and uncured cooked pork sausages. *Meat Sci.* **89**, 150-153.
- Jia, N., Kong, B., Liu, Q., Diao, X., and Xia, X. (2012) Antioxidant activity of black currant (*Ribes nigrum* L.) extract and its inhibitory effect on lipid and protein oxidation of pork patties. *Meat Sci.* **91**, 533-539.
- Jun, H. I., Kim, Y. A., and Kim, Y. S. (2014) Antioxidant activities of *Rubuscoreanus* Miquel and *Morus alba* L. fruits. *J. Korean Soc. Food Sci. Nutr.* **43**, 381-388.
- Kim, H. S., Kang, E. J., Kim, W. S., and Kim, M. H. (2014) Study to find the optimal purification processing conditions of anthocyanin from *Bokbunja* byproducts. *Food Eng. Prog.* **18**, 25-31.
- Kim, H. S., Kwon, T. W., Lee, Y. S., Choung, M. G., and Moon, G. S. (2005) A major antioxidative components and comparison of antioxidative activities in black soybean. *Korean J. Food Sci. Technol.* **37**, 73-77.
- Kim, M. J., Song, Y. J., Kim, H. R., Lee, S. R., Sok, D. E., Kim, S. N., and Kim, R. M. (2009) Polyphenol and phytate contents and their relationship to antioxidative activity in soybeans. *J. East Asian Soc. Dietary Life.* **19**, 975-980.
- Kim, N. M., Sung, H. S., and Kim, W. J. (1993) Effect of solvents and some extraction conditions on antioxidant activity in cinnamon extracts. *Korean J. Food Sci. Technol.* **25**, 204-209.
- Kim, S. Y., Ko, K. O., Lee, Y. S., Kim, H. S., and Kim, Y. H. (2008) Extraction efficiency and stability of anthocyanin pigments in black soybean seed coat. *Korean J. Crop Sci.* **53**, 84-88.
- Lee, E. J. and Ahn, D. U. (2005) Quality characteristics of irradiated turkey breast rolls formulated with plum extract. *Meat Sci.* **71**, 300-305.
- Lee, H. J., Choi, E. Y., Sim, Y. J., Kim, O. S., Yoo, H. J., Do, W. N., and Kim, Y. H. (2009) Anthocyanin-contents and pigment stability of black soybean by different extract condition and stabilizer. *J. Korean Soc. Food Sci. Nutr.* **22**, 150-157.
- Lee, H. J., Do, J. R., Jung, S. K., and Kim, H. K. (2014) Physiological properties of *Sarcodon aspratus* extracts by ethanol concentration. *J. Korean Soc. Food Sci. Nutr.* **43**, 656-660.
- Lee, J. J., Lee, J. S., Choi, Y. I., and Lee, H. J. (2013) Antioxidant activity of Sansa (*Crataegi fructus*) and its application to the pork tteokgalbi. *Korean J. Food Sci. Ani.* **33**, 531-541.
- Lee, K. J., Yun, I. J., Kim, H. Y., Kim, K. H., Kim, Y. J., Kim, D. W., and Lim, S. H. (2010) Antioxidative activity of solvent extracts from *Synurus excelsus* and *Synurus palmartopinnatifidus*. *J. Korean Soc. Food Sci. Nutr.* **39**, 1893-1897.
- Lee, S., Lee, Y. B., and Kim, H. S. (2013) Analysis of the general and functional components of various soybeans. *J. Korean Soc. Food Sci. Nutr.* **42**, 1255-1262.
- Min, B. and Ahn, D. U. (2005) Mechanism of lipid peroxidation in meat and meat products-A review. *Food Sci. Biotechnol.* **14**, 152-163.
- Myung, J. G. and Hwang, I. K. (2008) Functional components and antioxidative activities of soybean extracts. *Korea Soybean Digest.* **25**, 23-29.
- Naveena, B. M., Sen, A. R., Vaithyanathan, S., Badji, Y., and

- Kondaiah, N. (2008) Comparative efficacy of pomegranate juice, pomegranate rind powder extract and BHT as antioxidants in cooked chicken patties. *Meat Sci.* **80**, 1304-1308.
32. Neci, P. S. and Jayaprakasha, G. K. (2003) Antioxidant and antibacterial activities of *Punica granatum* peel extracts. *J. Food Sci.* **68**, 1473-1477.
33. Pietta, P. G. (2000) Flavonoids as antioxidants. *J. Nat. Prod.* **63**, 1035-1042.
34. Resurreccion, A. V. A. (2003) Sensory aspects of consumer choices for meat and meat products. *Meat Sci.* **66**, 11-20.
35. Sarma, A. D., Sreelakshmi, Y., and Shama, R. (1997) Antioxidant ability of anthocyanins against ascorbic acid oxidation. *Phytochemistry* **45**, 671-674.
36. Seydim, A. C., Guzel-Seydim, Z. B., Acton, J. C., and Dawson, P. L. (2006) Effects of rosemary extract and sodium lactate on quality of vacuum packaged ground ostrich meat. *J. Food Sci.* **71**, 71-76.
37. Shah, A. M., Bosco, S. J. D., and Mir, S. A. (2014) Plant extracts as natural antioxidants in meat and meat products. *Meat Sci.* **98**, 21-33.
38. Slinkard, K. and Singleton, V. K. (1977) Total phenol analysis: Automation and comparison with manual methods. *Am. J. Enol. Viticult.* **28**, 49-55.
39. Song, D. J., Jeong, T. C., Moon, J. D., Kim, Y. G., Shin, T. S., Lee, J. I., Park, T. S., and Park, G. B. (1997) Influence of ethanol extracted propolis on lipids oxidation with modified pH ground pork. *Korean J. Food Sci. An.* **17**, 131-139.
40. Takahashi, R., Ohmori, R., Kiyose, C., Momiyama, Y., Ohsumi, F., and Kondo, K. (2005) Antioxidant activities of black and yellow soybeans against low density lipoprotein oxidation. *J. Agric. Food Chem.* **53**, 4578-4582.
41. Takahata, Y., Ohnishi-kameyama, M., Furuta, S., Takahashi, M., and Suda, I. (2001) Highly polymerized procyanidins in brown soybean seed coat with a high radical-scavenging activity. *J. Agric. Food Chem.* **49**, 5843-5847.
42. Tarladgis, B. G., Watts, B. M., Younthan, M. T., and Dugan, L. R. (1960) A distillation method for the quantitative determination of malonaldehyde in rancid foods. *J. Am. Oil Chem. Soc.* **37**, 403-406.
43. Viskelis, P., Rubinskienė, M., Jasutienė, I., Šarkinas, A., Dabaravicius, R., and Česonienė, L. (2009) Anthocyanins, antioxidative, and antimicrobial properties of American cranberry (*Vaccinium macrocarpon* Ait.) and their press cakes. *J. Food Sci.* **74**, C157-C161.
44. Wrolstad, R. E., Durst, R. W., and Lee, J. M. (2005) Tracking color and pigment changes in anthocyanin products. *Food Sci. Technol.* **16**, 423-428.
45. Wu, A. H., Ziegler, R. G., Nomura, A. M., West, D. W., Kolonel, L. N., Horn-Ross, P. L., Hoover, R. N., and Pike, M. C. (1998) Soy intake and risk of breast cancer in Asians and Asian Americans. *Am. J. Clin. Nutr.* **68**, 14375-14435.
46. Youn, J. T., Kim, J. T., Park, H. J., Kim, S. L., Kim, S. J., Kwon, Y. U., Chung, I. M., and Kim, M. J. (2014) Quality characteristics of soybean cured processed with green cotyledon colored soybean. *Korean J. Int. Agric.* **26**, 135-140.